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# A brief review on Triazin-pyridazinones: Synthesis and biological activities

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**Abstract:** A series of substituted triazin-pyridazine compounds were exhibited diverse types of biological activities and synthesized by different methods. These compounds were mainly tested for their anti-inflammatory, anticancer, antifungal and antibacterial activities. These substituted triazin-pyridazine compounds have mild to potent activities on compare with their appropriate reference standards.

Keywords: Triazolo-pyridazinones, Cytotoxicity, Pyridazin, antifungal, antibacterial, triazole

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#### INTRODUCTION

Heterocyclic annelated pyridazines attract considerable attention, which mainly arises from the large variety of interesting pharmacological activities observed with pyridazine derivatives [1-6]. In recent years, synthesis of novel pyridazinone derivatives and investigation of their chemical and biological behavior have gained more importance due to their biological, medicinal, and agricultural reason. This privileged structure attracts the interest of medicinal chemists as a nucleus of potential therapeutic utility and exhibits several pharmacological activities such as analgesic, anti-inflammatory, antidepressant, antihypertensive, anticonvulsant, cardiotonic, diuretic, anticancer, and anti-HIV activities [7-12]. In addition, pyridazinones act as core nucleus in various drugs e.g. Sul-mazole, Amipizone, Indolidan, Levosimendan, Imazodan, Pimobendan, Emorfazone, Zardaverine and Milrinone [13-15]. Triazole and its derivatives are noteworthy for their physiological and biological importance. They paved the attention of medicinal chemist due to their wide range of biological activities like anti-inflammatory, antiviral, antifungal, antibacterial, anticonvulsant and anticancerousetc [16-20]. Triazolopyridazine derivatives are frequently used in biological research [21,22].

Anti inflammatory and analgesic activity: Non-steroidal anti-inflammatory drugs (NSAIDs) are the most commonly prescribed medications for the treatment and/or management of pain, fever, and inflammation. However, their long term use is linked with gastroenteropathy such as gastric irritation, ulceration, bleeding and renal toxicity that limit their therapeutic usefulness [23]. Therefore, the synthesis of new compounds devoid of such side effects poses a challenging task for medicinal chemists. The currently used NSAIDs inhibit the synthesis of non

selective or selective cyclooxygenases (COX1 & COX2 or prostaglandin, histamine and bradykinin [24-26]. It has been suggested that prostaglandins and bradykinins play a major role in the analgesia or pain. So it may be pre dicted that the title compounds may act by inhibiting the synthesis of these chemical mediators involved in causing pain as well as inflammation [27]. The pyridazinones having characteristic pharmacological features, relative stability and ease of preparation contemplated us to synthesize some new derivatives of pyridazinones as good analgesic and anti-inflammatory agents. Various substituted pyridazinone derivatives possessing analgesic activity along with other useful pharmacological properties have been reported. Emorfazone (4-ethoxy-2-methyl-5-morpholino-3(2H)-pyridazinone) is a pyridazinone derivative which is currently used clinically in the management of pain and inflammation. Antinociceptive activity of 4-amino-2-methyl-6-phenyl-5-vinyl-3(2H)-pyridazinonewhich was found to be many fold potent than the emorfazone. A series of 2/3-substituted-6(4-methylphenyl)-4,5-dihydro-pyridazin3(2H)-ones and Pyridazine substituted Triazine were developed that allow efficient preparation of analogues with varied substitutions on the pyridazinone ring [28]. Somepyridazine derivatives were evaluation of their analgesic and anti-inflammatory activities to obtainesafer non-steroidal anti-inflammatory agents (NSAIDs). The aryl propionic acid on reaction with phenyl hydrazine and hydrazine hydrate yielded the pyridazinone derivative, 6-(4-Methylphenyl)-2-phenylpyridazin-3(2H)-one (1) and 6-(4-Methylphenyl)-4,5-dihydropyridazin-3(2H)-one respectively. The reaction of the compound 2 with phosphorus oxychloride (POCI2) produced the corresponding chloropyridazine derivative, 3-Chloro-6-(4-methylphenyl)-pyridazine (3). A 2-(Hydroxymethyl)-6-(4-methylphenyl)-4,5-dihydropyridazin-3(2H)-one (4)was synthesized by condensing 2 with methanol and formaldehyde (HCHO). The compound 4 on further treatment with

guanidine hydrochloride in ethanol gave the pyridazino-triazine, 7-(4-methylphenyl)-3,4,8,9-tetrahydro-2H-pyridazino[1,6-á][1,3,5]triazin-2-imine (5). The compounds
were tested for their analgesic activity in mice and
anti-inflammatory activity in Wistar albino rats. The results
of *in-vivo* anti-inflammatory studies revealed that the compound 3 showed maximum inhibition in paw edema volume
followed by compound 2 while the compound 3 exhibited
excellent peripheral analgesic activity (74%) followed of
the compound 4. Compounds 3 and 4 also showed a good
central analgesic effect increased the reaction time to 90
minutes [28].

Compounds, 2 and 3, had smaller paw volume than the positive control and were found to be more potent. All the tested compounds except compound 4 exhibited the maximum reduction in paw volume at 120 min while 4 displayed the maximum protection at 90 min. It could be concluded that the anti-inflammatory activity of the synthesized compounds could be due to the inhibition of inflammatory mediators release and possibly due to the inhibition of COX synthesis similar to indomethacin. It was observed that substitution of 2-phenyl ring at p-position with -CH<sub>2</sub>OH in compound 4 decreases the anti-inflammatory activity. However, the best activity is shown by compound (2) having no substituent at second position of pyridazine ring. Further, replacing an oxo group at third position with a -chloro group does not lead to change in the activity. Compound (3) was found to be the most potent analgesic agent with respect to standard drug. Other compounds also showed moderate to good analgesic activity. The results indicated that pyridazine derivatives possess significant analgesic activity associated with NSAIDs properties. All title compounds exhibited anti-inflammatory activity that lasted for 90 min and the potency increased with time. Among the synthesized pyridazinones, compound 3 emerged as lead compound with good analgesic and anti-inflammatory activities at par with the reference drug. Most of the compounds were exhibited and analgesic effect by both peripheral and central mechanisms. The anti-inflammatory and analgesic potential of 2/3 substituted pyridazine derivatives was confirmed. However, further detailed investigations are needed to establish the safety, efficacy and mechanism of this promising class of heterocyclic compounds [28]. Antimicrobial activity: A series of 3-substituted phenyl-6-substituted phenyl(1,2,4)triazolo(4,3-b) pyridazine has been synthesized. The corresponding aroyl propionic acid was cyclised with hydrazine hydrate to give 6-(substituted aryl)-2,3,4,5-tetrahydro-3-pyridazinone, which was heated on steam bath with phosphorous oxy chloride to

yield 3-chloro-6-substituted phenyl pyridazine. This intermediate after reaction with hydrazine hydrate was converted into 3-hydrazino-6-substituted phenyl pyradazine. The resulting product was converted into 3-substituted phenyl 6-substituted phenyl(1,2,4)triazolo(4,3-b)pyridazine by reacting with substituted aroyl chloride. The synthesized compounds were investigated for their in vitro antifungal and antibacterial activities. The results indicated that the synthesized compounds have mild to potent activities with reference to their appropriate reference standards [29]. The synthesized 3-phenyl 6-(4'-tolyl) (1,2,4) triazolo (4,3b) pyridazine (**6a**), 3,6-di(4'-tolyl) (1,2,4)triazolo(4,3-b) pyridazine (6b), 3-phenyl 6-(2',4'-dimethyl phenyl)(1,2,4) triazolo(4,3-b) pyridazine (7a), 3-(4'-tolyl)-6-(2',4'-dimethyl phenyl)(1,2,4) triazolo(4,3-b)pyridazine (7b), 3-phenyl6-(4'-phenoxyphenyl)(1,2,4)triazolo(4,3-b)pyridazine (8a).3-(4'-tolyl)-6-(4'-phenoxyphenyl)(1,2,4)triazolo(4,3-b) pyridazine (8b), 3-phenyl 6-(4'-chloro phenyl)(1,2,4) triazolo(4,3-b)pyridazine (9a), 3-(4'-tolyl)-6-(4'-chloro phenyl)(1,2,4) triazolo(4,3-b) pyridazine (9b), 3-phenyl-6-(4'-bromo phenyl)(1,2,4)triazolo(4,3-b)pyridazine (**10a**) and 3-(4'-tolyl)-6-(4'-bromo phenyl)(1,2,4)triazolo (4,3-b) pyridazine (10b) compounds were evaluated for their antibacterial activity against EscherichiaColi, Staphylococcus aureus, Micrococcus luteus and Klebsiella pneumonia. The results of antibacterial exhibit that all compounds having comparable activity against the bacterial strain. Compounds 6b, 7b and 8b are the most active derivatives, which shows significant activity against these bacteria comparable to standard drug, ampicillin and chloramphenicol. All the final compounds were evaluated for antifungal activity against Candidaalbicans and Candidaneoformans and compared with standard drug fluconazole. The compounds 7a, 7b and 9b were found to be active derivatives of this series against the microorganism. It is concluded that compounds 6b, 7b and 8b are active against gram positive and gram negative bacteria. Compound 7a, 7b and 9b are potent antifungal drugs [29]. The 6-substituted phenyl-2-[{(4'-substituted phenyl-5'-thioxo)-1,2,4-triazol-3-yl}-methyl]-2,3,4,5tetrahydropyridazin-3-one (11-20) compounds were tested for their in vitro antifungal activity on five fungal species, namely Candida albicans, Trichophyton rubrum, Aspergillus flavus, Aspergillus niger and Penicillium citrinium. The antifungal activities of the 6-Substituted-2-{[(4'substituted-phenyl-5íthioxo)-1,2,4-triazol-3-yl]-methyl]-2,3,4,5-tetrahydropyridazin-3-one compounds (11-20) against different Candida albicans, Trichophyton rubrum, Aspergillus flavus, Aspergillus niger and Penicillium citrinium fungal species and the results were compared with the standard drug voriconazole. The MIC of voriconazole for all the fungal species was lower than 0.5µg/mL. All compounds were found significant antifungal activities against all the fungal species. The chloro substituent derivative (17) showed the highest activity against all the fungal species. The MIC of the voriconazole was between 0.10 and 0.50 µg/ml against all the fungal species. The two electronegative groups of CI was increasing the activity of 1,2,4-triazole. The bulky group or aromatic group on benzene ring decreased the activity.

**Anticancer activity:** The triazolo[4,3-b]pyridazinones were evaluated for their potential *in vitro* cytotoxic

antitumor properties. The compounds were prepared by 1,3-dipolar cyclo addition of pyridazin-3-ones with N-aryl-C-ethoxycarbonylnitrile imines, generated *in situ* from ethylhydrazono- $\alpha$ -bromoglyoxylates.

The triazolo[4,3-b]pyridazinones (21-27) have been

synthesized with the aim to evaluate their cytotoxic activity against tumor cells. Cells (105/ml) were treated with increasing concentrations of compounds **22-24**, followed by 48 h incubation. *In vitro* evaluation of the cytotoxic effect of compounds **23** and **24** showed that both of these

Scheme 1. Synthesis of 2-amino-2-(6-chloro-[1,2,4]triazolo[4,3-b]pyridazin-3-yl) acetic acid from ethyl 2-(benzamido)-2-(6-chloro-[1,2,4]triazolo[4,3-b]pyridazin-3-yl) acetate.

Scheme 2. Synthesis of 2-amino-2-(6-chloro-[1,2,4]triazolo[4,3-b]pyridazin-3-yl) acetic acid and N-((6-hydroxy-[1,2,4] triazolo[4,3-b]pyridazin-3-yl)methyl) benzamide from ethyl 2-(benzamido)-2-(6-chloro-[1,2,4]triazolo[4,3-b]pyridazin-3-yl) acetate.

Scheme 3. Synthesis of 2-((6-chloro-[1,2,4]triazolo[4,3-b]pyridazin-3-yl)methyl) isoindoline-1,3-dione and (6-chloro-[1,2,4]triazolo[4,3-b]pyridazin-3-yl) methanamine

Scheme 4. Synthesis of 2-(benzamido)-2-(6-chloro-[1,2,4]triazolo[4,3-b]pyridazin-3-yl) acetic acid, N-((6-chloro-[1,2,4]triazolo[4,3-b]pyridazin-3-yl)methyl) benzamide and compound (XI) from ethyl 2-(benzamido)-2-(6-chloro-[1,2,4]triazolo[4,3-b]pyridazin-3-yl) acetate (I).

Scheme 5. Preparation of N-Benzoyl-(6-hydrazino[1,2,4] triazolo [4,3,-b] pyridazin -3-yl) glycine hydrazide and 2-Hydroxyethyl N-benzoyl –(6-chloro [1,2,4] triazolo [4,3-b] pyridazin-3-yl) glycinate

products are cytotoxic to the HepA cell line in a dose-dependent manner, with  $IC_{50}$  values of 20.76 µg ml-1 and 9.2 µg ml-1, respectively. It is noteworthy that the cytotoxic effect of these products is weaker than that induced by adriamycin ( $IC_{50}$ :1.2 µg ml-1). The compound **24** was more cytotoxic than compound **23**. Compound **24** exhibited significant cytotoxicity against the Hep cell line [31].

**Synthesis:** The preparation of some derivatives of the [1,2,4]triazolo[4,3-b]pyridazine system from ethyl N-benzoyl-(6-chloro[1,2,4]triazolo[4,3-b]pyridazin-3-yl)glycinate is reported [32]. A general approach to ethyl *N*-benzoyl- $\alpha$ -hetero-aryl-glycinates based on the annulation of the 1,2,4-triazole ring on the suitably substituted azine or fused azine derivative (**Scheme 1**). The [1,2,4]triazolo[4,3-b]pyridazine system has attracted great attention, it explore synthetic utility of compound I in the preparation of some other derivatives. In particular, interest in the preparation of the amino acid II [21].

Reaction of I with hot 1 M sodium hydroxide solution took place on the ethoxycarbonyl group and at the position 6 of the triazolopyridazine system to give III, which underwent decarboxylation in hot ethanol affording benzamide IV (Scheme 2) [21].

Refluxing of I in diluted hydrochloric acid (1:1) for 4 h gave hydrochloride V. This compound was also prepared via a longer reaction sequence starting from hydrazide VI which was firstly transformed by heating in acetic acid into the chloromethyl substituted triazolopyridazine VII. Substitution of the chloro group with potassium phthalimide in hot N,N-dimethylformamide produced VIII, which hydrolyzed in hot hydrochloric acid to give V (Scheme 3). On the other hand, treatment of I with concentrated hydrochloric acid at room temperature for 15 days yielded acid IX, which similarly as III underwent decarboxylation by heating in ethanol giving benzamide X. The latter was also prepared by oxidative cyclization of hydrazone XI [34]. Evidently, strong reaction conditions required for the elimination of the protective groups caused the decarboxylation of IX making impossible the preparation of the acid II (Scheme 4).

Stable triazolopyridazine were obtained in reactions of 1 with hydrazine hydrate and ethylene glycol. Reaction with hot 80% hydrazine hydrate proceeded similarly as with sodium hydroxide at two electron deficient carbons, carbon 6 and carbonyl group, giving hydrazide XII, whereas treatment with hot ethylene glycol resulted in the formation of XIII. Both products might be suitable starting compounds for the preparation of some tricyclic systems, the former also for the peptide synthesis (Scheme 5).

### CONCLUSION

Pyridazine is an important heterocyclic scaffold for designing medicinal agents with varying biological actions. The easy functionalization at various ring positions makes them an attractive synthetic building block for designing, synthesis and discovery of new drugs. The incorporation of this versatile biologically accepted pharmacophore in established medicinally active molecules results in wide range of pharmacological effects. Further optimization of the chemical synthesis can possibly lead to more active molecules against fungal infections. Since all twelve compounds showed promising results, studies to establish

their in vivo efficacy will be carried in the future.

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